PHYTOCHEMICAL REPORTS

CAROTENOIDS OF SOUTHERN PINE RUST CRONARTIUM FUSIFORME

L. R. G. VALADON and DAVID PORTER

Department of Botany, Royal Holloway College, Englefield Green, Surrey TW20 OEX, England, and Department of Botany, University of Georgia, Athens, GA 30602, U.S.A.

(Received 30 August 1973. Accepted 20 September 1973)

Key Word Index—*Cronartium fusiforme*; Fungi; pine rust; aeciospores; carotenoids; β -carotene; lycopene.

Large quantities of aeciospores of *Cronartium fusiforme* Hedg. and Hunt are found on loblolly pine (*Pinus taeda* L.) in the Athens area in April and the carotenoids of this fungus were investigated.

The following carotenoids were identified in the spore material before hydrolysis; β -carotene (34·8%), γ -carotene (63·1%) and lycopene (2·1%) in 161·7 μ g/g dry wt total carotenoids. After harsh "hydrolysis" (see Experimental), the following carotenoids were observed: β -carotene (18·2%), β -zeacarotene (2·6%), ζ -carotene (1·9%), γ -carotene (64·6%), isomer β -carotene? (8·4%), lycopene (2·3%), zeaxanthin (0·3%), flavoxanthin (1·1%) and auroxanthin (0·6%) out of a total of 485·5 μ g/g dry wt.

Our results, expressed as a percentage of total carotenoids, before and after hydrolysis showed that the percentages of γ -carotene and lycopene were not altered but that of β -carotene was reduced by almost half, with the appearance of β -zea- and ζ -carotenes. The presence of small amounts of xanthophylls in the fraction after hydrolysis was unexpected. The presence of 5,8-furanoid oxides, namely auroxanthin and flavoxanthin, is not known in organisms that lack chloroplasts. It is most likely, therefore, that xanthophylls were obtained from minute contaminants and not from the spores themselves.

EXPERIMENTAL

The aeciospores of C. fusiforme were collected from infected pines by brushing the spores from the galls into a polyethylene bag. The spores were then passed through a 100 mesh sieve to eliminate the larger debris, stored at -36° , and extracted within a week after collection. The extraction was carried out in a Virtis homogenizer $2 \times \text{MeOH}$, $2 \times \text{MeOH}$ -Et₂O(1:1, v/v) and then $4 \times \text{with}$ acetone at which time no more colour was extracted. This treatment is sufficient to extract all the carotenoids from most fungi; however, in this case, the spores were still brightly coloured. Irvine et al. found that if they digested their rust spores with N HCl for 1 hr at 100° , the residual pigments could be easily extracted. This we carried out, followed by centrifugation and extraction of the spore material with Et₂O until no more colour came out. A large amount of carotenoids was released by this method, and the spore material was now dark brown. Two extracts were therefore obtained, one before and the other after hydrolysis.

Partition with Et₂O was performed and all the colour was transferred to the epiphasic layer after addition of H₂O. The upper layer which contained all the extracted pigments was concentrated under reduced pressure

¹ IRVINE, G. N., GOLUBCHUK. M. and ANDERSON, J. A. (1954) Can. J. Agric. Sci. 34, 234.

² HOUGEN, F. W., CRAIG, B. M. and LEDINGHAM, C. A. (1958) Can. J. Microbiol. 4, 521.

³ Bush, L. (1967) Phytopathology 57, 785.

⁴ WALKINSHAW, C. H. and SCHELD, H. W. (1973) in press.

at about 35°. Procedures for the phase-partition of carotenoids between hexane and aq. 90°_{\circ} (v/v) MeOH and the separation and identification of carotenoids by column chromatography and TLC were as described. 5.6

Concentrations of individual carotenoids were estimated and results were calculated on a per g dry-wt basis. The dry wt of the spore material was 1.5 g.

Acknowledgements—We would like to thank Arnold Foudin for collecting the large quantities of aeciospores. L.R.G.V. gratefully acknowledges financial support from the University of Georgia through a Postdoctoral Research Associateship.

- ⁵ Herber, R., Maudinas, B. and Villoutreix, J. (1972) Compt. Rend. 274, 327.
- 6 VALADON, L. R. G. and MUMMERY, R. S. (1968) Biochem. J. 106, 479.
- ⁷ Valadon, L. R. G. and Mummery, R. S. (1967) Ann. Bot. N.S. 31, 495.

Phytochemistry, 1974, Vol. 13, p. 650. Pergamon Press. Printed in England.

CHRYSOPHANOL AND EMODIN FROM DRECHSLERA CATENARIA

G. W. VAN EIJK

Centraalbureau voor Schimmelcultures, Baarn. The Netherlands

(Received 30 August 1973. Accepted 20 September 1973)

Key Word Index—*Drechslera catenaria*; Moniliales: Fungi; anthraquinones; chryophanol; emodin.

Plant. Drechslera catenaria (Drechsler) S. Ito CBS 191·29 = Helminthosporium catenarium. Previous work. Isolation and characterization of catenaria and helminthosporin from D. catenaria; 1·2 chrysophanol and emodin from different organisms, including fungi. 3

Present work. D. catenaria was grown at 24° on Czapek. Dox medium for 10 weeks. The dried mycelium (12.5 g) was finely ground and extracted successively with petrol and EtOAc. The residue (2.7 g) after the evaporation of EtOAc extract yielded 0.6 g helminthosporin and 1.7 g catenarin by means of fractional crystallization from EtOH. The mother liquor was concentrated in vacuo and the residue taken up in EtOAc. Neutral and acidic EtOAc fractions were prepared by means of the conventional method of 500 NA₂CO₃ separation. The neutral fraction (150 mg) was submitted to chromatography on silica gel G layers. The minor band was extracted with CHCl₃ leaving yellow leaflets (5 mg) m.p. $197-199^\circ$ after concentration. (MS Found: $254\cdot059235$. Calc. for $C_{15}H_{10}O_4$: 254·057902). This led to a tentative identification of the compound as chrysophanol. Comparison with an authentic sample (m.m.p., UV, IR, MS, TLC) established its identity. The acidic fraction (270 mg) yielded a further small amount of catenarin (220 mg) after concentration. TLC of the residue revealed a yellow spot with nearly the same R_f value as catenarin. Good separation on thin layer plates only succeeded via the acetates. Hydrolysis of the individual acetates gave catenarin and a small amount of orange needles (1 mg) m.p. 259–261°. (MS Found: 270-053650. Calc. for $C_{15}H_{10}O_5$: 270-052817). The compound proved to be emodin by direct comparison with a commercial sample (Fluka) (m.m.p., UV. IR. TLC).

Acknowledgements—The author thanks Mr. H. J. Roeymans for his technical assistance and Mr. C. Versluys, Analytical Laboratory, State University of Utrecht for measuring the MS.

¹ RAISTRICK, H., ROBINSON, R. and TODD, A. R. (1934) Biochem. J. 28, 559.

² Anslow, W. K. and Raistrick, H. (1940) Biochem. J. 34, 1124.

³ Thomson, R. H. (1971) Naturally Occurring Quinones, 2nd Edn., pp. 388, 419, Academic Press, London.